

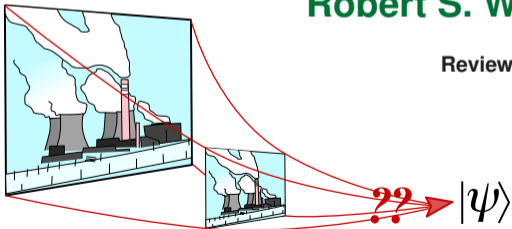


Laboratoire de Physique et Modélisation des Milieux Condensés
Univ. Grenoble Alpes & CNRS, Grenoble, France

CLOSED questions in heat transport and conversion at the nanoscale

Robert S. Whitney

Review – Benenti, Casati, Saito, R.W.
Physics Reports **694**, 1 (2017)



OVERVIEW

(I) Laws of thermodynamics same in quantum

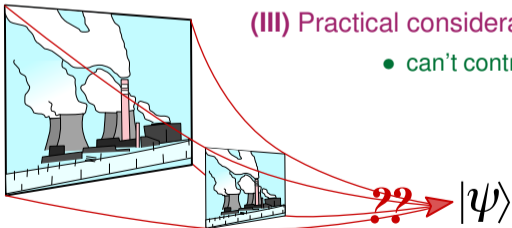
- where does entropy come from?
Schrodinger Eq. = Reversible
- Carnot efficiency

(II) Extra constraints from quantum

- quantized heat flow

(III) Practical considerations

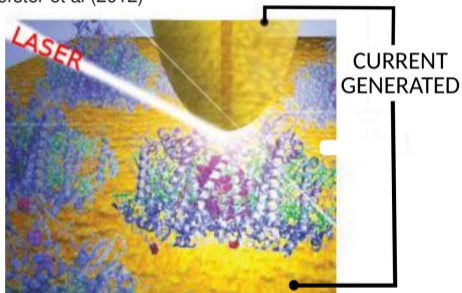
- can't control heat!



NANOSCALE MACHINES

Photosynthetic molecule (experiment)

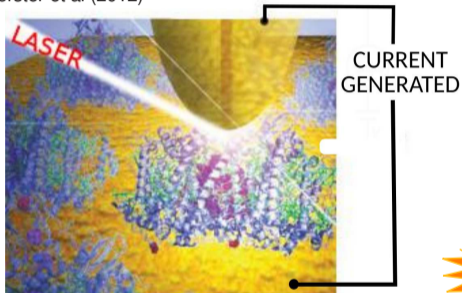
Gerster et al (2012)



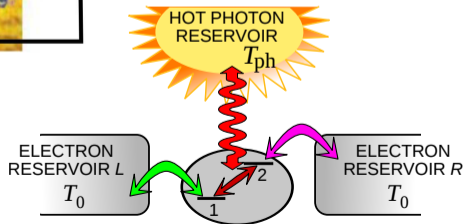
NANOSCALE MACHINES

Photosynthetic molecule (experiment)

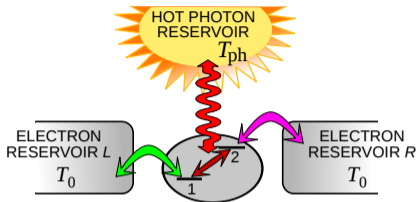
Gerster et al (2012)



SIMPLE MODEL



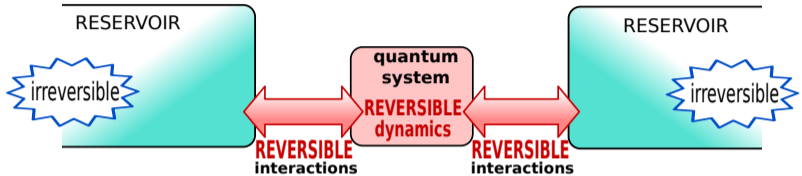
NANOSCALE MACHINES



FOR THIS TALK, I ASSUME:

- ♣ Nanoscale/quantum system between *macroscopic* reservoirs
 - Long times (steady state) \Rightarrow “macroscopic” work output
cf. fluctuation theorems
- ♣ Thermal reservoir states:
 - No squeezed reservoir states cf. J. Roßnagel, et al PRL (2014)
 - No non-equil. reservoir states
cf. Sanchez, Splettstoesser & Whitney "Demon preprint" (2018)

REVERSIBLE OR NOT

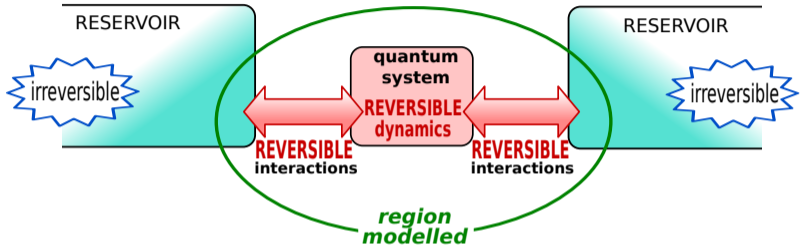


REVERSIBLE or NOT?

- Where is *entropy produced*?
- 2nd law of thermodynamics?



REVERSIBLE OR NOT

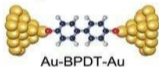
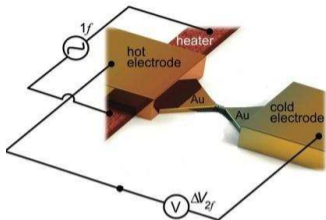


REVERSIBLE or NOT?

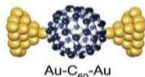
- Where is *entropy produced*?
- 2nd law of thermodynamics?



SIMPLER – NANOSCALE THERMOELECTRIC



Reddy group
(2015)



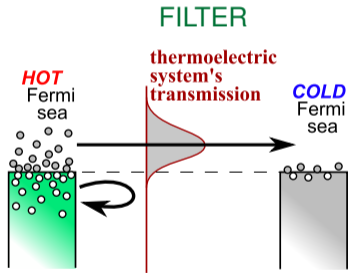
SIMPLE MODEL



SIMPLER – NANOSCALE THERMOELECTRIC

Acts as Energy FILTER

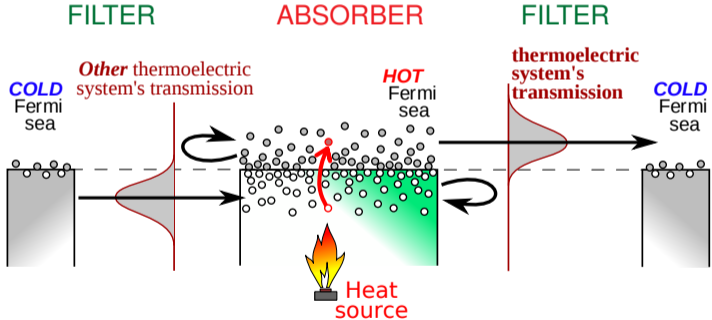
different dynamics *above* and *below* Fermi surface



SIMPLER – NANOSCALE THERMOELECTRIC

Acts as Energy FILTER

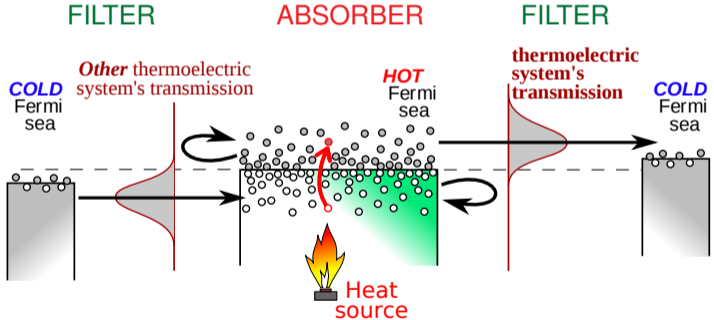
different dynamics *above* and *below* Fermi surface



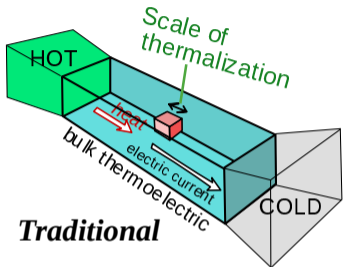
NANOSCALE THERMOELECTRIC

Acts as Energy FILTER

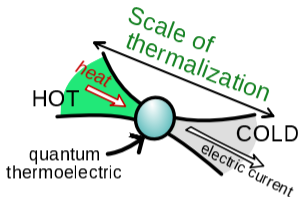
different dynamics *above* and *below* Fermi surface



TRADITIONAL versus QUANTUM



Traditional

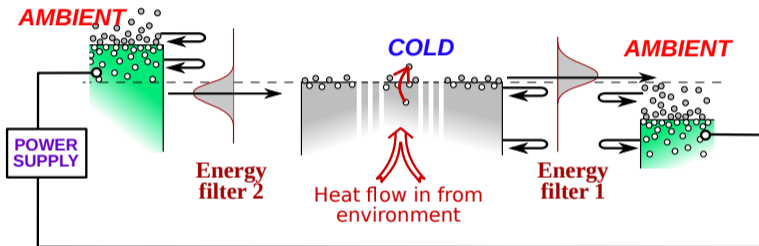


Quantum

- (1) Filters \ll scale of thermalisation \Rightarrow Quantum thermoelectric
- (2) Central region (absorber) also \ll scale of thermalisation \Rightarrow Quantum thermocouple
- Often filters-absorber *combined* in quantum thermocouples
cf. simple model of photosynthetic molecule

QUANTUM THERMOELECTRICS FOR REFRIGERATOR

PELTIER FRIDGE



EFFICIENCIES AND ZT

HEAT-ENGINE efficiency

$$\eta_{\text{engine}} = \frac{\text{work done}}{\text{heat used}}$$

Carnot limit

$$\eta_{\text{engine}}^{\text{Carnot}} = 1 - T_{\text{cold}}/T_{\text{hot}}$$

LINEAR RESPONSE: $\eta_{\text{engine}} = \eta_{\text{engine}}^{\text{Carnot}} \times \frac{\sqrt{ZT+1} - 1}{\sqrt{ZT+1} + 1}$

$$ZT = \frac{(\text{Electric Conductance}) (\text{Seebeck})^2 T}{\text{Thermal Conductance}}$$

$$\left\{ \begin{array}{l} ZT = \infty \Rightarrow \text{Carnot} \\ ZT = 3 \Rightarrow \text{Carnot}/3 \\ ZT = 1 \Rightarrow \text{Carnot}/6 \end{array} \right.$$

EFFICIENCIES AND ZT

HEAT-ENGINE efficiency

$$\eta_{\text{engine}} = \frac{\text{work done}}{\text{heat used}}$$

Carnot limit

$$\eta_{\text{engine}}^{\text{Carnot}} = 1 - T_{\text{cold}}/T_{\text{hot}}$$

LINEAR RESPONSE: $\eta_{\text{engine}} = \eta_{\text{engine}}^{\text{Carnot}} \times \frac{\sqrt{ZT+1} - 1}{\sqrt{ZT+1} + 1}$

$$ZT = \frac{(\text{Electric Conductance}) (\text{Seebeck})^2 T}{\text{Thermal Conductance}} \quad \left\{ \begin{array}{l} ZT = \infty \Rightarrow \text{Carnot} \\ ZT = 3 \Rightarrow \text{Carnot}/3 \\ ZT = 1 \Rightarrow \text{Carnot}/6 \end{array} \right.$$

FRIDGE efficiency

$$\eta_{\text{fridge}} = \frac{\text{heat removed}}{\text{work used}}$$

Carnot limit

$$\eta_{\text{fridge}}^{\text{Carnot}} = T_{\text{cold}}/(T_{\text{ambient}} - T_{\text{cold}})$$

"LINEAR" RESPONSE: Max T difference : $\frac{\Delta T}{T} \leq \frac{1}{2} ZT$

MODELLING METHODS

- Landauer scattering theory \Leftarrow *No e-e Interactions*

see chapters 4-6 of Benenti *et al*, Physics Reports **694**, 1 (2017)

- Rate Equations \Leftarrow *Weak Coupling*

see chapters 8-9 of Benenti *et al*, Physics Reports **694**, 1 (2017)

- Keldysh field theory \Leftarrow *“everything”* often need approx similar to above

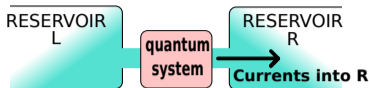
OPEN questions ask Rosa & David, or Fabienne & Adeline ...

- Schrodinger eq. for system+environment
-

LANDAUER SCATTERING THEORY

LANDAUER SCATTERING THEORY

(TWO TERMINALS)



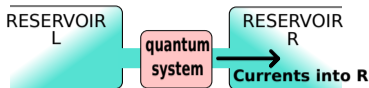
$$\text{Particle current into R} \equiv \frac{d}{dt} N_R = \int dE \mathcal{T}(E) \left[f\left(\frac{E-\mu_L}{T_L}\right) - f\left(\frac{E-\mu_R}{T_R}\right) \right]$$

$$\text{Energy current into R} \equiv \frac{d}{dt} E_R = \int dE E \mathcal{T}(E) \left[f\left(\frac{E-\mu_L}{T_L}\right) - f\left(\frac{E-\mu_R}{T_R}\right) \right]$$

UNITS: $\hbar = k_B = 1$

LANDAUER SCATTERING THEORY

(TWO TERMINALS)



$$\text{Particle current into R} \equiv \frac{d}{dt} N_R = \int dE \mathcal{T}(E) \left[f\left(\frac{E-\mu_L}{T_L}\right) - f\left(\frac{E-\mu_R}{T_R}\right) \right]$$

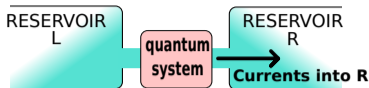
$$\text{Energy current into R} \equiv \frac{d}{dt} E_R = \int dE E \mathcal{T}(E) \left[f\left(\frac{E-\mu_L}{T_L}\right) - f\left(\frac{E-\mu_R}{T_R}\right) \right]$$

UNITS: $\hbar = k_B = 1$

- Electric current into R = $e \frac{d}{dt} N_R$
- Work into R = $\mu_R \frac{d}{dt} N_R \implies$ power generated = sum L & R
 $= (\mu_R - \mu_L) \frac{d}{dt} N_R$
- Heat current into R = $\frac{d}{dt} E_R - \mu_R \frac{d}{dt} N_R$
- Entropy current into R = Heat current / $T_R \leftarrow$ CLAUSIUS LAW (1855)

LANDAUER SCATTERING THEORY

(TWO TERMINALS)



The Blackboard

see section 6.4 of Phys. Rep. **694**, 1 (2017)

- Rate of change of entropy $\frac{d}{dt}S$

$$= (\text{Entropy current into L}) + (\text{Entropy current into R}) \geq 0$$

\Rightarrow 2nd law proven for all $\mathcal{T}(E), T_L, T_R, \dots$

Nenciu (2007) = proof for N reservoirs

- Carnot efficient system : reversible $\frac{d}{dt}S = 0$

requires transmission is only non-zero at $\frac{E-\mu_L}{T_L} = \frac{E-\mu_R}{T_R}$

Humphrey, Newbury, Taylor, Linke (2002)

Quantum bound on heat flow

Maximum HEAT FLOW per transverse mode \Rightarrow maximum ENTROPY CHANGE

Bekenstein, Phys. Rev. Lett. (1981)

Pendry, J. Phys. A (1983)

Heat current out of hot reservoir

$$J_{\text{hot}} \leq N \frac{\pi^2 k_B^2}{6h} (T_{\text{hot}}^2 - T_{\text{cold}}^2)$$

with “quantum” $N \sim \frac{\text{cross-section}}{\text{wavelength}}$



Quantum bound on heat flow

Maximum HEAT FLOW per transverse mode \Rightarrow maximum ENTROPY CHANGE

Bekenstein, Phys. Rev. Lett. (1981)

Pendry, J. Phys. A (1983)

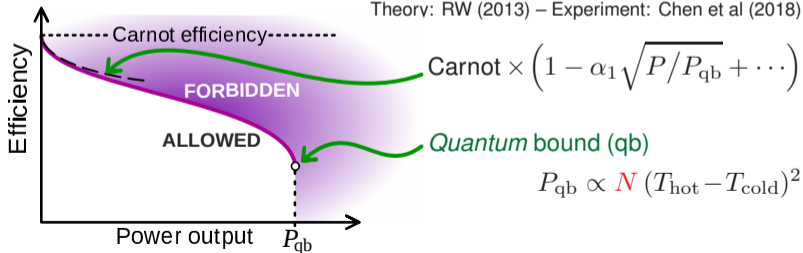
Heat current out of hot reservoir

$$J_{\text{hot}} \leq N \frac{\pi^2 k_B^2}{6h} (T_{\text{hot}}^2 - T_{\text{cold}}^2)$$

with "quantum" $N \sim \frac{\text{cross-section}}{\text{wavelength}}$

Quantum bound on power output

Theory: RW (2013) – Experiment: Chen et al (2018)



...do better with B-field/interactions cf. Brandner et al 2015, Lao et al 2018

Is quantum bound relevant for REAL applications?



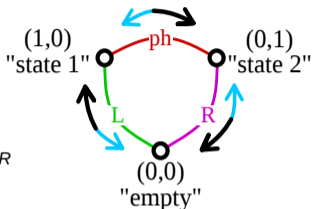
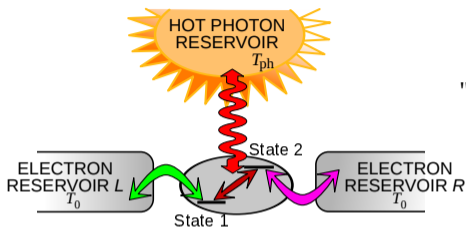
Cross-section for $100W$ of power ?

with wavelength $\lambda_F \sim 10^{-8}m$

- ◇ Minimal cross-section $\sim 4mm^2$
- ◇ 90% of Carnot requires $> 0.4cm^2$

RATE EQUATIONS

RATE EQUATIONS



Set of coupled equations for system evolution:

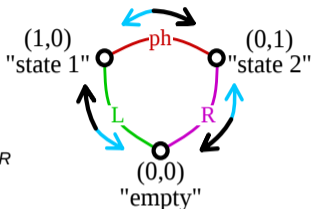
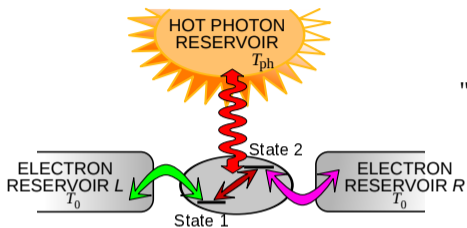
$$\frac{d}{dt}P_b(t) = \sum_a \left(\Gamma_{b \leftarrow a} P_a(t) - \Gamma_{a \leftarrow b} P_b(t) \right)$$

where P_b = prob. system is in state b

& $\Gamma_{b \leftarrow a}$ = rate $a \rightarrow b$

$$\text{Current into R} = \Gamma_{(0,0) \leftarrow (0,1)} P_{(0,1)} - \Gamma_{(0,1) \leftarrow (0,0)} P_{(0,0)}$$

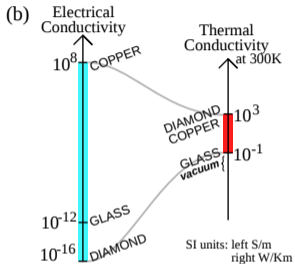
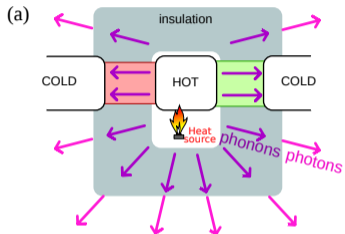
RATE EQUATIONS



USEFUL TRICKS – look at loops:

- only Carnot efficient if no loop generates entropy
- if single-loop system $\Rightarrow \eta_{\text{engine}} = \frac{\text{Work in one cycle}}{\text{Heat in one cycle}}$

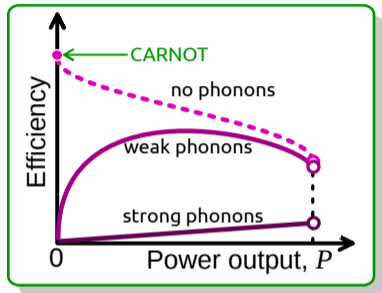
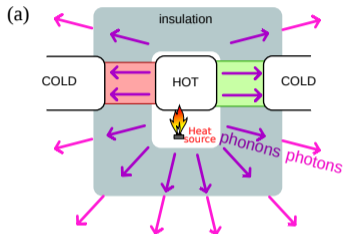
CAN'T CONTROL HEAT !!



Heat engine efficiency

$$\eta_{\text{engine}} = \frac{\text{work done}}{\text{heat used} + \text{heat lost}}$$

CAN'T CONTROL HEAT !!



Heat engine efficiency

$$\eta_{\text{engine}} = \frac{\text{work done}}{\text{heat used} + \text{heat lost}}$$

CONCLUSION: CLOSED[†] QUESTIONS

[†] *closed* for theories simple enough to treat
≡ (a) Landauer & (b) Rate eqs.

(I) Laws of thermodynamics same at nanoscale
if reservoirs are macroscopic & internal equilibrium

- Entropy produced
- Carnot efficiency (in principle) ← vanishing power

(II) Extra constraints from quantum

- quantized heat flow

(III) Practical considerations

- can't control heat!

